



**PROCESS TO OBTAIN POLYGLYCOLYL UREA FROM AROMATIC
DIGLYCINATES FOR ELECTRIC CONDUCTOR INSULATION**

BACKGROUND OF THE INVENTION

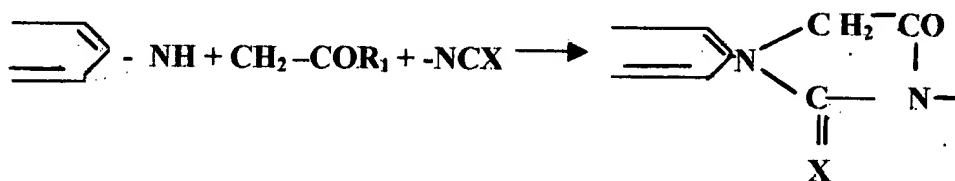
Several processes are known to obtain polyhydantoin polymers, that is, polymers containing hydantoin rings.

Among the known processes, we can mention the ones involving the reaction, through heating, of glycine derivatives and polyisocyanate or polythioisocyanates, at temperatures ranging from 80°C to 500°C, in presence of a solvent. In US patent No. 3,397,253, a process is described to obtain polyhydantoin polymers through heating glycine derivatives and polyisocyanates or polythioisocyanates at temperatures ranging from 80° C to 500° C, being said compounds preferably N,N'-bis-carbetoxy methyl-4,4' diisocyanate-diphenylmethane in a cresol solvent.

Another type of known processes ^{COMPRISING} ~~consists in~~ ^{ARYL} condensing ~~ary~~ polyamines with sodium or potassium cyanides and oxo compounds, for example, formaldehyde, ketone, etc. through the addition of acids; the nitriles obtained are saponified to form carboxylic acids or esters with hydrochloric acid in alcohol.

Other processes ^{COMPRISE} ~~consist~~ in modifying already prepared glycine derivatives, for example, through free acid sterification or ester aminolysis.

In this type of processes, the polymerization reaction is usually in addition to the condensation of the two reagents, ^{PRODUCING} ~~occurring~~ a ring closure reaction to form a hydantoin ring according to the following scheme:



US patent No. 5,006,641 teaches how to obtain polyhydantoins through a process involving the initial reaction of bis-glycine esters with diisocyanates in a liquid medium to form polyurea and then, through reaction, in a second stage, with a monoglycine ester equivalent to the residual quantity of isocyanate terminal groups and finally polyhydantoin is obtained through cycle condensation.

In US patent No. 5,041,518, a process is described to obtain polyhydantoins useful in coating materials, films, adhesives and molding through the reaction of polyfunctional carboxylic ^{ALPHA} amino acids with organic poly(thio)cyanates and the addition of carboxylic acid compounds in ^{THE} presence of condensed aromatic hydrocarbon at temperatures ranging from 0° C to 50° C.

The polymer compounds of hydantoin rings have several known applications according to their polymerization grade. When they are overheated at high temperatures, they turn into non melting high hardness products; it is also possible to obtain flexible coatings when polymerization is stopped before its final state and thus abrasion resistance and conventional resistance to several types of solvents can be obtained. Together with other shellac or mica products, materials in solutions as insulating elements for the electrical industry, especially electric coil wires, can be obtained.

SYN *BL* *INS A* *THE PRESENT INVENTION IS DIRECTED TO* (PGU)
~~The applicant has developed a process to obtain polyglycolyl urea from aromatic diglycinates, the main characteristic of which is that it does not form high risk polluting residual by-products such as HCN emission, obtaining a product that meets the main properties of such commercially available resins such as thermal, MECHANICAL and chemical properties and even improving certain characteristics such as freon resistance of polyesterimide-type enameled products.~~

INS A *2* *ANOTHER ASPECT*
~~The process object of the present invention is also characterized because it reduces the number of unitary operations conducted, ^{THUS LOWERING} ~~lowering thus~~ the cost of the product.~~

DESCRIPTION OF THE INVENTION

Hereinafter the invention will be described according to the process stages to obtain the polyglycolyl urea as well as its use, main object of the application, in the manufacturing of H-class magnet-wire with improved properties.

The process to obtain polyglycolyl urea PGU is divided in two main stages: A and B.

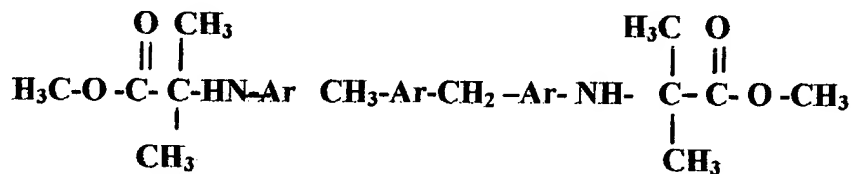
Stage A includes the following steps:

- 1) mixing the reaction solvents, bromided ester, diamine and catalyst;
- 2) Reacting the mixture to produce diglycinate in solution;
- 3) Purifying the diglycinate obtained through filtration, washing and drying;

Stage B includes the following steps:

- 4) loading aromatic isocyanate, diglycinate, solvents and catalyst in the polymerization reactor;
- 5) obtaining polyglycolyl urea resin;
- 6) adding polyester-type electro-insulating varnishes;
- 7) manufacturing H-class magnet-wire with improved properties.

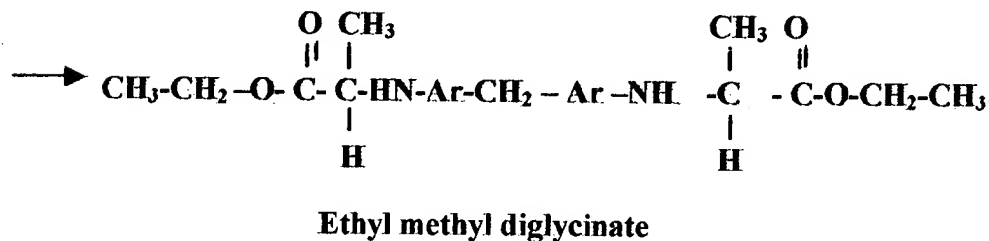
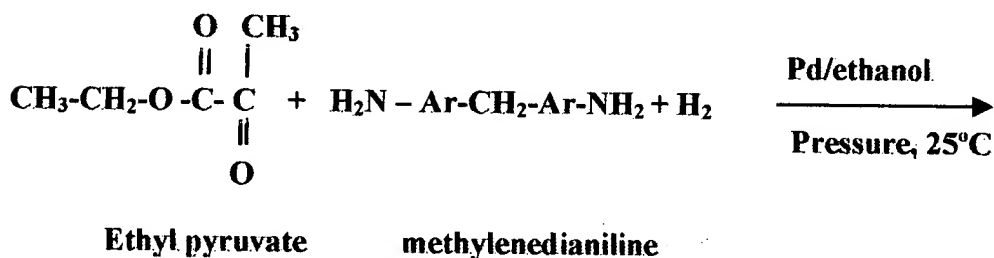
The chemical structure of the methyldimethyldiglycinate required to obtain the identical commercial polyhydantoin resin (PH-10 from Bayer A. G.) is as follows:



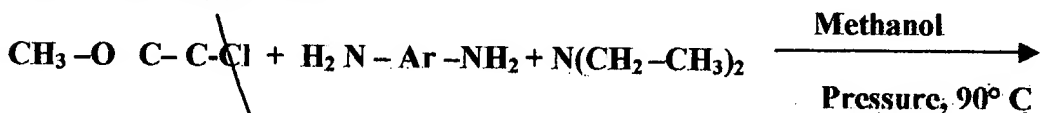
where Ar represents aromatic rings.

In order to develop a diglycinate countertype through an easier path without deriving dangerous, polluting by-products such as HCN and without presenting problems related to freon resistance when added with varnishes, as is the case with polyhydantoin resin, the applicant has identified two paths to obtain polyglycolyl urea:

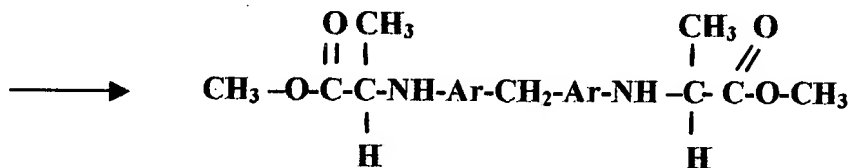
Path 1: reductive amination



Path 2: nucleophilic shift



Methylchloro- methylene- triethyl-
Propionate dianiline amine



Methyl methyl diglycinate

The nucleophilic shift process produces a diglycinate with better characteristics because it defines both methyl methyl diglycinate isomers.

Hereinafter Stage 2 is described, in which the halogen "chlorine" is substituted by "bromine" from the methyl propionate that gives better results in the removal of residual by-products.

STAGE A

Obtainment of methyl diglycinate from methylenedianiline

- a) in a glass or stainless steel matrass, provided with stirring means, reflux column, heating and cooling systems, the following materials are added: methylenedianiline, methanol and methyl bromopropionate.
- b) Under stirring, the solution is heated at methanol reflux temperature (58 – 63° C); once this temperature is reached, triethylamine addition starts, at a rate of 0.178 l/hr per Kg of product during a 3 – 5 hour period.
- c) The reaction is maintained under reflux during 19 hours.
- d) Start methanol distillation at atmospheric pressure till 40% of its initial volume is recovered; the distilled methanol contains methyl acrylate obtained as side by-product induced by the base.
- e) Cooling at 20° C under stirring (products begin to crystallize at 50° C with a slight viscosity increase), then add the drinking water volume adequate to dissolve the bromine salt obtained. Continue stirring during 0.5 hour to homogenize the suspension formed.
- f) Filter the suspension, letting it drain thoroughly in order to assure the removal of polar impurities with help from residual methanol; said impurities are identified in the mother waters as an oily residue that does not contain appreciable quantities of methyl diglycinate.

- A

Then the mother waters from step f), a mixture of reaction residues and by-products such as hydrated bromotriethylamine salts, are neutralized with sodium hydroxide and submitted to a secondary distillation to obtain sodium bromide in solution and 90% triethylamine.

i) PGU synthesis is conducted in a stainless steel reactor provided with stirring, thermal oil heating, condenser and cooling system.

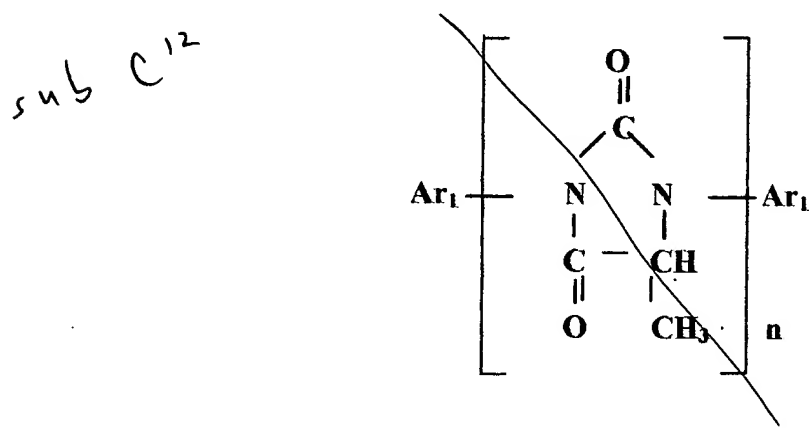
Cresylic acid and methyl methyl diglycinate are added to the reactor at room temperature, and the suspension is stirred till the solution of the product is complete.

- j) Then, methylene diisocyanate is added, under constant stirring and preventing the temperature from rising above 60° C, because this stage of the reaction is exothermic. Said conditions are maintained during 3 hours and then viscosity is determined in a No. 4 Ford Cup on a polymer sample, when the viscosity is within a range from 44 to 47 seconds at a temperature of 25° C.

- k) Then a triethylenediamine or 1,4-diazobicyclo (2,2,2) octane (DABCO) **CATALYST** is added, and the operation conditions are maintained during 1.5 hours. Once this period has elapsed, temperature is raised to 180 – 185°C in

about 6 hours. Once this temperature is reached, it is maintained till the specified quantity of distillate is obtained, during a period of about 5 to 6 hours and temperature can be increased up to 200°C, if necessary.

Once the theoretical distillate is recovered, heating is stopped and the resin is cooled at 70° C to unload the corresponding containers, and a polyglycolyl urea hydantoin resin of the following formula is obtained:



where Ar₁ is a substituted aromatic compound or a substituted diphenylalkyl, and 2 < n < 500, % solids = 28.97

Viscosity (Cp) = 4,800 at 15% solids.

In the following example the polyglycolyl urea (PGU) countertype resin is compared to a conventional resin PH-10 in a varnish.

Each resin is adjusted to a 1600 cps viscosity with cresol/xylol and ^{CATALYST} catalyst to be used as enameling varnishes. First, PH-10 was coated on conductor gauge 14 AWG to determine the work parameter to compare; a flexibility failure was observed at 6m/min. The results are presented in a parameter from 8 to 16 vel. (m/min).

PH-10 BASED VARNISH

Vel (m/min)	6	10	14	16
Adh. and Flex.	failure	good	good	good

Thermoplastic F.	271°	295°	316°	275°
Voltage (Kv)	9.0	9.19	7.61	5.69
Dissipation factor	2.9	2.90	18.1	56.9

PGU-BASED VARNISH

Vel (m/min)	8	10	12	14	16
Adh. and Flex.	good	good	good	good	97%
Thermal shock	good	good	good	good	93%
Thermoplastic E.	340°	310°	284°	249°	231°
Voltage (Kv)	7.43	7.80	7.34	8.44	7.04
Dissipation Factor	7.68	10.2	16.9	34.3	67.6

[illegible]